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Wednesday, January 17th, 2024
Princeton University, Time TBA

Prof. Viktor Cybulskis



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High-Silica Pd-Based Zeolite Catalysts for Lean-burn Natural Gas Emissions Control

Large natural gas engines, such as those used for power generation and maritime transportation, generate less CO₂ than coal or petroleum but the global warming impact of uncombusted methane (CH₄) emitted in the exhaust far exceeds CO₂ and must be abated by using catalytic aftertreatment devices. Although metal oxide-supported Pd catalysts are active for CH₄ oxidation, they require temperatures above 500 °C and high Pd loadings to convert CH₄ in the presence of combustion products such as H₂O, CO₂, SO₂. Such demanding operating conditions lead to fast and irreversible catalyst deactivation, hindering widespread commercial implementation. Zeolites are promising support materials to overcome these challenges due to their hydrothermal stability, resistance to dealumination in the presence of steam, and unique ability to confine active metals to prevent thermal sintering. Our recent work demonstrates that small- and medium-pore zeolites, such as CHA and MFI, respectively, with very high Si/Al ratios (50-∞) can stabilize reactive Pd

nanoparticles, either through encapsulation or metal-support anchoring, and increase CH₄ oxidation rates at low temperatures in the presence of H₂O. CH₄ oxidation performance after simulated aging at 650 °C for 1 h under wet-lean conditions (0.15% CH₄, 5% O₂, 5% H₂O, bal. Ar) reveals that 1 wt.% Pd/CHA catalysts with Si/Al > 33 attain > 90% CH₄ at temperatures below 400 °C compared to 500 °C for a conventional 1 wt.% Pd/Al₂O₃ catalyst. Furthermore, steady state CH₄ turnover frequencies over Pd-based CHA and MFI catalysts are up to 4× greater than 1 wt.% Pd/Al₂O₃, suggest that these high-silica zeolites contain Pd sites that more easily activate CH₄ by inhibiting the accumulation of inactive hydroxyl species on the Pd surface in the presence of H₂O while facilitating O₂ adsorption and transport to oxidize adsorbed CH₄. Increasing the zeolite Si/Al ratio leads to greater hydrophobicity and improved catalyst durability in presence of H₂O during CH₄ oxidation under lean-burn conditions by decreasing the mobility of active Pd species and reducing the inhibiting effect of H₂O on catalytic turnovers.

Speaker Bio

Viktor J. Cybulskis is an Assistant Professor in the Department of Biomedical and Chemical Engineering at Syracuse University. He received his B.S. in Chemical Engineering from Purdue University in 2005. He began an industrial career with LyondellBasell Industries, holding various roles in manufacturing and research and development for fuels and petrochemicals, and then joined TPC Group in 2009. He returned to Purdue University and received his Ph.D. in 2016 under the direction of Fabio H. Ribeiro and W. Nicholas Delgass. He was a postdoctoral scholar at the California Institute of Technology with Mark E. Davis. In August 2018, he joined the faculty at Syracuse University. His research focuses on understanding the molecular details of heterogeneously catalyzed reactions and designing reactive micro-environments to enable new pathways for selective chemical transformations. He received the ACS PRF Doctoral New Investigator Award (2022) and was selected as a finalist for the Beckman Young Investigator Award (2023). He is a licensed professional engineer, currently serves as Secretary of AIChE's Licensing and Professional Development Committee, and is regularly involved in session organization and chairing at national meetings for AIChE, ACS, and NAM. In his free time, he enjoys the outdoors, reading, watching sports, and spending time with his family.

Schedule TBA

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Deadline for reservations is 12:00PM Wednesday, January 10th, 2024

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